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(54) Title: NON- $\beta$ -OXIDIZABLE FATTY ACID ANALOGUES, THEIR USES AS THERAPEUTIC ACTIVE MEDICAMENTS, AND PREPARATION THEREOF			
(57) Abstract <p>There are disclosed compounds of the general formula (I): alkyl-X-CH<sub>2</sub> COOR wherein alkyl represents a saturated or unsaturated hydrocarbon of 8-26 carbon atoms, X represents a sulfur atom or a selenium atom and R is hydrogen or C<sub>1</sub>-C<sub>4</sub> alkyl. Said compounds are used for the manufacturing of medicaments for the treatment of hyperlipidemic conditions, (arteriosclerotic disease), coronary artery disease and for reducing the concentration of lipids in blood of mammals, for inhibiting oxidative modification of LDL, and for reducing proliferation of cancer cells. Methods for preparing the compounds are also disclosed.</p>			

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**NON- $\beta$ -OXIDIZABLE FATTY ACID ANALOGUES, THEIR USES AS  
THERAPEUTIC ACTIVE MEDICAMENTS, AND PREPARATION THEREOF.**

5 This invention relates to the use of certain non- $\beta$ -oxidizable fatty acid analogues for the manufacture of medicaments for the treatment of hypolipidemic conditions, such as for reducing the concentration of cholesterol and triglycerides, and for inhibition of the oxidative modification of low density 10 lipoprotein (LDL) in the blood of mammals. These medicaments also have a preventive effect on growth of tumour cells and may therefore be used in treatment of various terms of cancer diseases. The invention also relates to a method for preparing a medicament based on the mentioned fatty acid analogues, and 15 also relates to a new compound having all of the above given favourable therapeutical effects.

Excess of cholesterol and triglycerides in blood has been shown to accelerate the development of arteriosclerosis and 20 is a risk factor for myocardial infarction. Accordingly, a reduction of excess of lipids in blood by diets or drugs is used as a preventative measure in people at risk due to high levels of cholesterol and triglycerides and high platelet activation.

25 In this connection reference is made to European Patent Specification No. 345.038 (NORSK HYDRO A.S., priority of GB-8813012 of 2 June 1988) which discloses the use of non- $\beta$ -oxidizable fatty acid analogues of the general formula (I):

**Alkyl-X-CH<sub>2</sub> COOR**

wherein alkyl represents a saturated or unsaturated hydrocarbon group of from 8-22 carbon atoms, X represents O, S, SO, and SO<sub>2</sub>, and R represents hydrogen or C<sub>1</sub> - C<sub>4</sub> alkyl, for the manufacture of a medicament for the treatment of hyperlipaemic conditions and for reducing the concentration of cholesterol and triglycerides in the blood of mammals. The EP-specification also discloses the preparation of compounds of the actual non-β-oxidizable fatty acid analogues wherein the substituent X represents O, S, SO, SO<sub>2</sub>, respectively. The EP-specification reports that the compounds in question exhibit favourable hypolipidemic effects in blood of mammals, such as rats, and possess low toxicity measured as increase in liver weight and increased peroxisomal β-oxidation. The patent concludes that the compounds in question are potentially useful as medicinal compounds. For further considerations we refer to the EP-specification.

20

It has now been found that the analogues of the above mentioned non-β-oxidizable fatty acid have broader area of applications as an ingredient in drugs than the ones reported in the European Patent Specification No.

25

0.345.038. Further, it has been found that analogues with other election for the substituent X in the compound of formula (I), exhibit, as will be evident from the below specification, generally more potent pharmaceutical effects, also regarding the previously disclosed teaching of treatment of hypolipidemic conditions for reducing the concentration of cholesterol and triglycerides.

30

Current research suggests that prevention of atherosclerosis must take into consideration not just lowering plasma cholesterol and triglycerides, but also decreasing the susceptibility of LDL to oxidative damage. Oxidatively modified LDL - but not native LDL - has a number of characteristic properties that may initiate formation of foam cells and promote the development of fatty streaks, the

earliest lesions in atherosclerosis. It could therefore be expected that preventing LDL modification could alternate foam cell formation and the development of plaques.

It is well known that natural long-chain fatty acids,  
5 particularly polyunsaturated fatty acids of the main origin, are effective in lowering plasma triglyceride but not cholesterol levels in man. Moreover, as a lipid radical-propagated peroxidation chain reaction, in which the polyunsaturated fatty acids contained in the LDL are  
10 rapidly oxidized to lipid hydroperoxides, high supplementation of main based diets rich in  $\omega$ -3-fatty acids may rather increase LDL oxidation.

The antiatherogenic properties of probucol are supposed to  
15 be related to its antioxidant effect rather than to its relatively weak hypocholesterolemic potency.

New strategies have been developed to search for compounds that are likely to protect against radical damage in order  
20 to prevent modification of LDL and at the same time be an effective lipid-lowering drug. Considering that polyunsaturated fatty acids are metabolized slowly, we postulated that simple fatty acid analogues blocked for  $\beta$ -oxidation with a reducing agent or atom might result in a  
25 very potent compound being able to inhibit LDL oxidative modification and to lower blood lipids.

In feeding experiments with such fatty acid analogues the results show that they lower the blood concentration of  
30 cholesterol and triglycerides and inhibit LDL oxidative modification, without any overt toxic effect.

These fatty acids analogues are to the best of our knowledge the simplest lipid and antioxidant compounds found so far.  
35

Considering that there is substantial evidence that polyunsaturated fatty acids from the n-3 family ( $\omega$ -3 fatty

5 acids) which are metabolized relatively slowly and reduce proliferation of cancer cells, we postulated that simple non- $\beta$ -oxidizable fatty acid analogues might have similar effects. Results of in vitro experiments with such fatty  
acid analogues show that they reduce the rate of proliferation and effect differentiation of cancer cells much more effectively than pure  $\omega$ -3 fatty acids do.

10 Thus, the present invention provides fatty acid analogues with the ability a) to lower concentration of cholesterol and triglycerides in the blood, b) to inhibit LDL oxidative modification, and c) to reduce the rate of proliferation of cancer cells. The fatty acid analogues of the present  
15 invention provides improved effect relative to  $\omega$ -3 fatty acids and without undesirable side effects.

More particularly, the present invention relates to the use of non- $\beta$ -oxidizable fatty acid analogues of the general formula (I)

20 **Alkyl-X-CH<sub>2</sub>, COOR**

wherein alkyl represents a saturated or unsaturated hydrocarbon group of from 8-26 carbon atoms, X represents a selenium atom, and R is hydrogen or C<sub>1</sub> - C<sub>4</sub> alkyl, for the  
25 manufacture of a medicament for

- a) the treatment of hyperlipidemic and antiatherogenic conditions, such as for reducing the concentration of cholesterol and triglycerides in the blood of mammals,
- b) to inhibit the oxidative modification of low density lipoprotein (LDL),  
30 and
- c) to reduce the growth of cancer cells.

In accordance with another aspect of the invention use is  
35 made of non- $\beta$ -oxidizable fatty acid analogues compounds of the above mentioned formula (I) and definitions for Alkyl and R, but wherein the substituent X represents a sulfur

atom, in order to inhibit the oxidative modification of low density lipoprotein (LDL), and to reduce the growth of cancer cells.

5 Preferably use is made of compounds as defined in the following claims 3-5.

In accordance with a further aspect of the invention therapeutic active medicaments are manufactured in that a 10 compound in question is incorporated in a pharmaceutical acceptable carrier or diluent, as is defined in the following claims 6-8.

15 In accordance with yet another aspect of the invention the present invention comprises a fatty acid analogue of the general formula (I):

**Alkyl-X-CH<sub>2</sub> COOR**

20 wherein Alkyl represents a saturated or unsaturated hydrocarbon group of from 8-22 carbon atoms, X represents a selenium atom and R is hydrogen or C<sub>1</sub> - C<sub>4</sub> alkyl.

25 Preferably, the fatty acid analogue compound of the general formula (I) is as defined in any of the claims 3-5.

In the preferred embodiment of the present invention "alkyl" represents a tetradecyl group.

30 The compounds used according to the present invention wherein the substituent X is a sulphur atom or a selenium atom may be prepared according to the following general procedures:

X is a sulfur atom:

The thio-substituted compound used according to the present invention may be prepared by the general procedure indicated below:



The preparation of a number of non- $\beta$ -oxidizable fatty acid derivates of formula (I) above will now be given by way of illustration.

Example 1.Synthesis of Tetradecylthioacetic acid

$\text{CH}_3-(\text{CH}_2)_{13}-\text{S-CH}_2\text{-COOH}$ . Compound I.

KOH, 20 g (0,3 equivalents), mercaptoacetic acid, 12 ml (0,14 equivalents), and tetradecyl bromide, 25 ml (0,09 equivalents), were added in that order to 200 ml methanol and the solution was stirred overnight in a nitrogen atmosphere. A white precipitate of potassium bromide was formed. To the reaction mixture concentrated HCl (30 ml) dissolved in water (400 ml) was then added. Tetradecylthioacetic acid started to precipitate immediately and the solution was left overnight at room temperature to complete this process. The product was then isolated by filtration and washed four times with water. After drying the product was crystallized once from diethyl ether and then twice from methanol. Tetradecylthioacetic acid appeared as white flakes with a melting point of 68°C.

Yield: 23 g = 75% as based on the amount of tetradecyl bromide used.  $^1\text{H-NMR}$  (in  $\text{CDCl}_3$ )  $\delta$  0,84-0,91 (t, 3H,  $\text{CH}_3$ ), 1,25-1,45 (m, 22H, 11  $\text{CH}_2$ ) 1,60-1,73 (p, 2H, - $\text{CH}_2\text{CH}_2\text{S-}$ ), 2,62-2,66 (t, 4H, - $\text{CH}_2\text{S-}$ ), 3,24 (s, 2H, S- $\text{CH}_2\text{COOH}$ ), 10,6 (s, 1H,  $\text{COOH}$ ).

X is a selenium atom:

The seleno-substituted compound used according to the present invention may be prepared by the following general procedure:

- 5      1. Alkyl-Hal + KSeCN    ----> Alkyl-SeCN
2. Alkyl-SeCN + BH<sub>4</sub><sup>-</sup>    ----> Alkyl-Se<sup>-</sup>
3. Alkyl-Se<sup>-</sup> + O<sub>2</sub>    ----> Alkyl-Se-Se-Alkyl

10      This compound is purified by careful crystallization from ethanol or methanol.

4. Alkyl-Se-Se-Alkyl    -----> 2 Alkyl-Se<sup>-</sup>
5. Alkyl-Se<sup>-</sup> + Hal-CH<sub>2</sub>-COOH    ---> Alkyl-Se-CG<sub>2</sub>-COOH

15      The final compound, f.ex. when Alkyl is tetradecyl, CH<sub>3</sub>(CH<sub>2</sub>)<sub>13</sub>-Se-CH<sub>2</sub>-COOH, can be purified by crystallization from diethyl ether/hexane. This product may be fully characterized by NMR, IR and molecular weight determination.

20      In the present case, the selenium compound tetradecyl-selenoacetic acid of formula CH<sub>3</sub>(CH<sub>2</sub>)<sub>13</sub>-Se-CH<sub>2</sub>-COOH (Compound II) was prepared.

25      Example 2

Synthesis of tetradecylselenoacetic acid:

CH<sub>3</sub>-(CH<sub>2</sub>)<sub>13</sub>-Se-CH<sub>2</sub>-COOH. Compound II.

30      1. Synthesis of ditetradecyl diselenide: CH<sub>3</sub>(CH<sub>2</sub>)<sub>13</sub>-Se-Se-(CH<sub>2</sub>)<sub>13</sub>CH<sub>3</sub>.

Black selenium, 3,54 g (0,045 mol), was suspended in 150 ml 1:1 mixture of tetrahydrofuran (THF) and water in an argon atmosphere. Sodium borohydride (NaBH<sub>4</sub>), 3,93 g (0,10 mol), 35 in 60 ml of argon-flushed THF:H<sub>2</sub>O (1:1) was added dropwise to the suspension (careful, exothermic). A reddish brown colour was initially formed but gradually disappeared. To this solution was added selenium, 3,54 g (0,045 mol),

suspended in 150 ml THF:H<sub>2</sub>O (1:1). A reddish-brown solution was formed. The reaction mixture was stirred for 15 min and finally heated for about 10 min to complete the dissolution of selenium. Tetradearyl bromide, 24,9 g (0,09 mol), in THF (50 ml) was then added to the solution. During one hour with gentle heating the solution turned yellow indicating the reaction to be completed. The reaction mixture was treated with chloroform and the organic layer was dried over anhydrous magnesium sulphate, filtered and evaporated to leave a yellow oil solidified upon cooling.

Crystallization from diethyl ether gave yellow needles with a melting point of 43°C.

Yield: 20 g = 80% as based on the amount of tetradearyl bromide used.

15

2. Synthesis of tetradeylselenoacetic acid: CH<sub>3</sub>(CH<sub>2</sub>)<sub>13</sub>-Se-CH<sub>2</sub>COOH. Compound II.

To the diselenide, 1,0 g (0,0018 mol), in 25 ml THF (freshly distilled from benzophenone and sodium) was added 20 dripwise NaBH<sub>4</sub>, 0,206 g (0,0054 mol), dissolved in 10 ml H<sub>2</sub>O in an argon atmosphere. After the solution had been decolorized with bromoacetic acid 1,0 g (0,0072 mol), and triethylamine, 1,0 ml (0,0072 mol), in THF (25 ml) was added and the solution was stirred for 6 h at room 25 temperature. Water (50 ml) was then added and the solution was treated with diethyl ether. The organic layer was discarded. The aqueous layer was acidified with HCl and extracted with diethyl ether. The ether layer was dried over anhydrous magnesium sulphate, filtered and was 30 evaporated in vacuum leaving a white solid.

Crystallizations, one from hexane (30 ml) and then from diethyl ether (30 ml), left white crystals of tetradeylselenoacetic acid with a melting point of 68°C.

Yield: 0,80 g = 66% as based on the amount of ditetradeyl 35 diselenide used.

<sup>1</sup>H-NMR (in CDCl<sub>3</sub>): δ 0,84-0,91 (t, 3H, CH<sub>3</sub>), 1,25-1,45 (m, 22H, 11 CH<sub>2</sub>), 1,62-1,73 (p, 2H, -CH<sub>2</sub>Se-), 2,8-3,06 (t, 4H, -CH<sub>2</sub>Se-), 3,15 (s, 2H, -Se-CH<sub>2</sub>COOH), 10,6 (s, 1H, COOH).

5

The pharmaceutical effects of the compounds prepared as disclosed above according to the present invention will now be disclosed further in the following experiments which are presented in the tables. The compounds I and II prepared as disclosed above were used in the experiments.

10

### EXPERIMENTS

15

#### Hypolipidemic effect

Male wistar rats, weighing 180-200 g at the start of the experiment, were housed individually in metal wire cages in a room maintained at 12 h light-dark cycles and a constant temperature of 20±3 °C. The animals were acclimatized for 20 one week under these conditions before the start of the experiments.

25

Compound I (tetradecylthioacetic acid), compound II (tetradecylselenoacetic acid) prepared in accordance with Examples 1 and 2, and eicosapentaenoic acid (EPA) were suspended in 0,5% (w/v) carboxymethyl cellulose (CMC). Six animals were used for each treatment and a 0,5% CMC solution was administered to rats as control. After administration of the test compound, rats were fasted for 12 hours and anesthetized with haloethan. Eicosapentaenoic acid and the fatty acid derivatives were administered by gastric intubation (gavage) once daily for 7 days. Blood samples were collected by cardiac puncture, and lipid concentrations in plasma were determined using an autoanalyzer. Results obtained with Se-tetradecylselenoacetic acid (compound II), EPA and tetradecylthioacetic acid (compound I), are reported in Table 1.

30

35

Table 1: Effect of compound I, compound II and EPA - hypolipidemic drug on plasma lipid levels in rats.

	Compound	Dose mg/day/kg body weight	Decreased Plasma lipids (% of control)	
			triglycerides	cholesterol
10	Compound II (Se-tetradecyl- selenoacetic acid)	15	25	20
15	Eicosapenta- enoic acid	1500	20	18
20	Compound I (Tetradecylthio- acetic acid)	150	45	30

Table 1 shows that tetradecylselenoacetic acid (Compound II) exhibits a good hypolipidemic effect in blood of mammals, such as rats, and possesses low toxicity measured as increase in liver weight and increased peroxisomal  $\beta$ -oxidation (data not shown). It will appear that a 100 times greater dose of the hypolipidemic drug eicosapentaenoic acid is necessary to obtain the same decreased plasma lipid results as obtained for compound II (tetradecylselenoacetic acid). Moreover, the substituted fatty acid compounds are much more effective than pure EPA and fish oil in lowering plasma lipids. Therefore they are potentially useful as medical compounds.

In another set of experiments hepatocytes from rats, not treated with the test compounds, were prepared. Cultured hepatocytes were incubated for 4 hours with [ $1-^{14}\text{C}$ ] palmitic acid (200  $\mu\text{M}$ ) in the presence of L-carnitine (0,5 mM) and the different drugs (Table 2) and medium triglycerides (secreted) were extracted and dissolved in n-hexane and separated by thin-layer chromatography on silica plates developed in hexane-diethylether-glacial acetic acid with a ratio of 80:20:1. The bands were visualized by iodine vapor, cut into pieces and counted.

Table 2:

Effect of tetradecylthioacetic acid (compound I), tetradecylselenoacetic acid (compound II), the EPA and oleic acid hypolipidemic drugs (200  $\mu$ M) on secretion of triglyceride-labeled [ $1-^{14}\text{C}$ ] palmitic acid from hepatocytes incubated with [ $1-^{14}\text{C}$ ] palmitic acid (200  $\mu$ M). Results are given as mean  $\pm$  SD for values obtained from five independent experiments.

	Compound	Secretion of [ $1-^{14}\text{C}$ ] palmitic acid labeled triglycerides (nmol/protein/4 hours)
15	Compound II tetradecyl- selenoacetic acid	14,4 $\pm$ 6,7**
20	Eicosapenta- enoic acid	24,4 $\pm$ 10,2*
25	Compound I Tetradecylthio- acetic acid	18,4 $\pm$ 5,9**
	Oleic acid	34,7 $\pm$ 5,96

\*  $p < 0,05$  compared to oleic acid (control).

\*\*  $p < 0,01$  compared control.

Table 2 shows that hepatocytes of rats grown with tetradecylselenoacetic acid and tetradecylthioacetic acid caused a statistically significant lower secretion of palmitic-acid - labeled triglycerides than did oleic acid.

#### Antioxidant effect

#### Example 3.

Male wistar rats, weighing 180–200 g at the start of the experiment, were housed individually in metal wire cages in a room maintained at 12 h light-dark cycles and a constant temperature of  $20\pm3^\circ\text{C}$ . The animals were acclimatized for one week under these conditions before the start of the experiments.

Compounds I and II according to the invention, and other fatty acid derivatives, were suspended in 0,5% (w/v) carboxymethyl cellulose (CMC). The fatty acid derivatives were administered by gastric intubation (gavage) once daily 5 for different days. The antioxidant effect as a function of the dose administered was examined.

Six animals were used for each treatment and a 0,5% CMC solution was administered to rats as control. After 10 administration of the test compound, rats were fasted for 12 hours and anesthetized with haloethan. Blood samples were collected by cardiac puncture and LDL preparations were prepared by ultracentrifugation.

15 In other set of experiments where the acid derivatives were administered at a dose of 250 mg/day/kg body weight, the antioxidant effects of different fatty acid derivatives were compared with that of control. In this experiment the dosing lasted for 7 days. In all these in vivo experiments, 20 adding tetradecylselenoacetic acid (compound II) and tetradecylthioacetic acid (compound I) to plasma as an antioxidant to prevent modification of LDL, in vitro, dramatically increase the lag time (data not shown). Thus the results indicate that compounds I and II achieves a 25 modification of LDL as the lag time increased. Therefore they are potentially useful as medical antioxidants.

Example 4.

Low-density lipoproteins (LDL) were prepared from fresh 30 normal human plasma by sequential ultracentrifugation. LDL were taken as the 1.021 to 1.063 density fraction, dialysed and the oxidation was initiated by addition of CuSO<sub>4</sub>. The kinetics of LDL oxidation were determined by monitoring the change in absorbence at 234 nm (nano meter) (conjugated 35 dienes). The change in absorbence at 234 nm vs time could be divided into three consecutive phases: lag, propagation and decomposition where the lag time is defined as the interval (minutes) between the intercept of the linear

least-square slope of the curve with the initial-absorbence axis.

Table 3 shows that addition of tetradecylselenoacetic acid (compound II) and tetradecylthioacetic acid (compound I) increased the lag time in a dose-dependent manner of Cu<sup>2+</sup>-treated LDL. Tetradecylselenoacetic acid (compound II) was much more potent than tetradecylthioacetic acid in the same experimental conditions. Addition of palmitic acid analogues, oxidized 3-thia fatty acid and 3-oxygen substituted fatty acid analogues did not changed the modification of LDL (Table 3).

Table 3. Effect of tetradecylselenoacetic acid (compound II) and tetradecylthioacetic acid (compound I) on modification of LDL from human plasma.

Compound added/concentration	lag Time (min)
No addition (control)	44,1 ± 0,3
Palmitic acid 5 µM	48,2 ± 4,8
10 µM	51,6 ± 4,8
20 µM	50,6 ± 9,6
Tetradecylthioacetic acid (compound I):	
5 µM	58,1 ± 9,6
10 µM	78,5 ± 11,6
20 µM	111,3 ± 23,1

continuation of Table 3.

Tetradecylselenoacetic acid (Compound II):

	1 $\mu\text{M}$	74,2 $\pm$ 6,8*
	2 $\mu\text{M}$	86,9 $\pm$ 7,6*
5	3 $\mu\text{M}$	152,3 $\pm$ 11,5*

Tetradecylsulfonylacetic acid:

	20 $\mu\text{M}$	48,2 $\pm$ 7,1
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Tetradecyloxyacetic acid

10	20 $\mu\text{M}$	51,6 $\pm$ 6,2
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\* P < 0,05 compared to control

For such purposes, the compounds of the present invention  
can be administered orally or parenterally in a conventional dosage or parenterally in a conventional dosage form such as tablets, capsules, powders, emulsions and solutions prepared according to conventional pharmaceutical practices.

20

Reduced proliferation of cancer cell

The effect of tetradecylselenoacetic acid (compound II) and tetradecylthioacetic acid (compound I) on many different cell lines, as specified in the left column of the following table 4, were studied.

Generally, the cells were grown in a humidified (95%) atmosphere of 5 % CO<sub>2</sub> and 95 % O<sub>2</sub> (air) maintained at 37 °C. All experiments were carried out using cell culture plates initially, the cells were plated, i.e. allowed to attach to the bottom, by incubating with plating medium CDulbeccos minimum essential medium.

35

Each of the test compounds I and II was incubated to each of isolated cell populations in a concentration of 100  $\mu\text{M}$ . Palmitic acid in a concentration of 100  $\mu\text{M}$  was also added to a population as a control sample. The cell number in all

samples was counted six days after the start time of the incubation by using standard techniques.

For the cells of mamma cancer (MCF-7) also Eicosapentaenoic acid and Docahexaenoic acid were tested similarly in concentrations of 100 µM each.

The results are presented in table 4 as the cell number in each sample following six days of incubation in percentage of the control, wherein the palmitic acid (the control) is given as 100 %.

Table 4. Effect of tetradecylselenoacetic acid (compound II), tetradecylthioacetic acid (compound I), and unsaturated fatty acids on cancer cell growth.

	Cells	Compounds	Cell number following six days of incubation (% of control)
20	Brain glioma rat	Palmitic acid	100
		compound II	60
		Tetradecylthioacetic acid	65
25	BT4C	Palmitic acid	98
		compound II	70
		Tetradecylthioacetic acid	68
30	BT4CN	Palmitic acid	100
		compound II	40
		Tetradecylthioacetic acid	35
35	Brain glioma, human	Palmitic acid	100
		compound II	55
		Tetradecylthioacetic acid	50
40	D-37MG	Palmitic acid	100
		compound II	60
		Tetradecylthioacetic acid	55
45	D-54MG	Palmitic acid	100
		compound II	60
		Tetradecylthioacetic acid	55

continuation of Table 4.Leucemic,  
human

5	HL-60	Palmitic acid	100
		compound II	40
		Tetradecylthioacetic acid	35
10	KG1a	Palmitic acid	100
		compound II	60
		Tetradecylthioacetic acid	65
15	Mamma cancer MCF-7	Palmitic acid	100
		compound II	60
		Tetradecylthioacetic acid	55
		Eicosapentaenoic acid	76
		Docahexaenoic acid	98

20

The added concentration of the different fatty acids was 100  $\mu$ M.

25 For all cell lines each of the compounds I and II exhibits a significant lower value for the count of the cell numbers than do the control compound of palmitic acid. For most of the tested compounds a reduction of proliferation of up to 40% or more was obtained.

30

As also will appear from table 4, compounds I and II inhibit the proliferation of the cell line of mamma cancer MCF-7 to a greater extent than eicosapentaenoic acid and docahexenoic acid.

35

The effect of various doses of compound I, compound II, palmitic acid, eicosapentaenoic acid and docahexenoic acid, on the cell number was measured by adding each of said compounds at different concentrations, i.e. at

40 concentrations of 10, 20, 50, 100, and 150  $\mu$ M, to isolated cell cultures of MCF-7 breast cancer. The number of cancer cells were counted 6 days following incubation by using standard techniques. The results are shown in the following table 5.

45

Table 5. Effect of tetradecylselenoacetic acid and tetradecylthioacetic acid at different doses on MCF-7 breast cancer cell growth.

	Compound	Cell numbers ( $\times 10^3$ ) in the presence of fatty acids at different concentrations ( $\mu\text{M}$ )					
		0	10	20	50	100	150
10	Palmitic acid	746	658	715	642	710	730
15	Tetradecylseleno-acetic acid	737	689	590	520	440	420
20	Tetradecylthio-acetic acid	740	637	570	480	410	400
25	Eicosapentaenoic acid	630	665	605	520	480	420
	Docahexaenoic acid	583	615	626	624	599	610

Table 5 shows that the compounds to various extents effected a reduction in cell number, i.e. a cancer cell proliferation. However, the proliferation was significantly greater for compounds I and II than for the other compounds in the test.

Tables 4 and 5 show that compounds I and II achieved a significant reduction in the rate of proliferation of cancer cells. Therefore the compounds are potentially useful as medicinal compounds.

The compounds used according to the present invention may be administered to patients suffering from any type of dyslipidaemia except type I. As antioxidants they can be used for various cardiovascular diseases. Regarding the reduced proliferation of cancer cells, they may be administered to patients suffering from any type of cancer. Alternatively by dietary they may prevent disease as atherosclerosis and tumor formation.

The dosage range for the compounds according to the present application is contemplated to be from 5 to 100 mg/day for the average adult patient. Of course, the actual dose necessary will depend on the patient's condition and will have to be determined by the attending physician from case-to-case.

For oral pharmaceutical compositions such carrier material as, for example, water, gelatine, gums, lactose, starches, magnesium-stearate, talc, oils, polyalkylene glycol, petroleum jelly and the like may be used. Such pharmaceutical preparation may be in unit dosage form and may additionally contain other therapeutically valuable substances or conventional pharmaceutical adjuvants such as preservatives, stabilizing agents, emulsifiers, buffers and the like. The pharmaceutical preparations may be in conventional solid dosage forms such as tablets, capsules, dragees and the like, in conventional liquid forms such as solutions, suspension, emulsions and the like, and other conventional dosage forms such as dry ampulles, suppositories and the like.

For parenteral administration the compounds according to the present invention may be administered as solutions, suspensions or emulsions using conventional pharmaceutical carrier materials such as for example water for injection, oils, polyalkylene glycols and the like. These pharmaceutical preparations may further include conventional pharmaceutical adjuvants, such as preservatives, stabilizing agents, wetting agents, emulsifiers, salts for the adjustment of the osmotic pressure, buffers and the like. The preparations may also contain other therapeutically active materials.

CLAIMS

5 1. Use of non- $\beta$ -oxidizable fatty acid analogues of the general formula (I)



10 wherein alkyl represents a saturated or unsaturated hydrocarbon group of from 8-26 carbon atoms, X represents a selenium atom, and R is hydrogen or C<sub>1</sub> - C<sub>4</sub> alkyl, for the manufacture of a medicament:

15 a) for the treatment of hyperlipidemic and antiatherogenic conditions, such as for reducing the concentration of cholesterol and triglycerides in the blood of mammals,  
b) to inhibit the oxidative modification of low density lipoprotein (LDL), and  
c) to reduce the growth of cancer cells

20 2. Use of non- $\beta$ -oxidizable fatty acid analogues of the general formula (I)



25 wherein alkyl represents a saturated or unsaturated hydrocarbon group of from 8-26 carbon atoms, X represents a sulfur atom, and R is hydrogen or C<sub>1</sub> - C<sub>4</sub> alkyl, for the manufacture of a medicament

30 a) to inhibit the oxidative modification of low density lipoprotein (LDL), and  
b) to reduce the growth of cancer cells.

3. Use according to claim 1 or 2, wherein alkyl represents the tetradecyl group.

4. Use according to claim 2, wherein the compound of formula (I) is tetradecylthioacetic acid.

5 5. Use according to any of claims 1 - 3, wherein the compound of formula (I) is tetradecylselenoacetic acid.

10 6. A process for the manufacture of a medicament for a) treatment of hypolipaemic conditions and for reducing the concentration of cholesterol and triglycerides in the blood of mammals, b) to inhibit the oxidative modification of low density lipoprotein (LDL), and c) to reduce the growth of cancer cells,

15 comprising incorporating with a pharmaceutical acceptable carrier or diluent, a non- $\beta$ -oxidizable fatty acid analogue of the general formula (I):

**Alkyl-X-CH<sub>2</sub> COOR**

20 wherein Alkyl represents a saturated or unsaturated hydrocarbon group of from 8-22 carbon atoms, X represents a selenium atom and R is hydrogen of C<sub>1</sub> - C<sub>4</sub> alkyl.

25 7. A process for the manufacture of a medicament to inhibit the oxidative modification of low density lipoprotein (LDL), and to reduce the growth of cancer cells,

comprising incorporating with a pharmaceutical acceptable carrier or diluent, a non- $\beta$ -oxidizable fatty acid analogue of the general formula (I):

30

**Alkyl-X-CH<sub>2</sub> COOR**

wherein Alkyl represents a saturated or unsaturated hydrocarbon group of from 8-22 carbon atoms, X represents a sulfur atom and R is hydrogen or C<sub>1</sub> - C<sub>4</sub> alkyl.

8. A process according to Claim 6 or 7, wherein the compound of general formula (I) is as defined in any of Claims 3-5.

5

9. Fatty acid analogue of the general formula (I):

**Alkyl-X-CH<sub>2</sub> COOR**

10 wherein Alkyl represents a saturated or unsaturated hydrocarbon group of from 8-22 carbon atoms, X represents a selenium atom and R is hydrogen of C<sub>1</sub> - C<sub>4</sub> alkyl.

15 10. Fatty acid analogue according to Claim 9 wherein the compound of general formula (I) is as defined in any of Claims 3-5.

## INTERNATIONAL SEARCH REPORT

International application No.

PCT/NO 95/00195

## A. CLASSIFICATION OF SUBJECT MATTER

**IPC6: A61K 31/19**

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

**IPC6: A61K**

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

**SE,DK,FI,NO classes as above**

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	STN International, File CA, Chemical Abstracts, volume 124, no. 9, 26 February 1996, (Columbus Ohio, US), Froeyland, Livar et al: "Tetradecyl-thioacetic acid incorporated into very low density lipoprotein: changes in the fatty acid composition and reduced plasma lipids in cholesterol-fed hamsters", abstract no. 115981, & J. Lipid Res. (1995), 36 (12), 2529-40 --	1-10
X	EP 0345038 A2 (NORSK HYDRO A.S.), 6 December 1989 (06.12.89) -----	1,3-10

 Further documents are listed in the continuation of Box C. See patent family annex.

\* Special categories of cited documents:

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- "E" earlier document but published on or after the international filing date
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- "&" document member of the same patent family

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**INTERNATIONAL SEARCH REPORT**  
Information on patent family members

01/10/96

International application No.  
**PCT/NO 95/00195**

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
EP-A2- 0345038	06/12/89	SE-T3- 0345038 CA-A- 1329550 DE-D,T- 68910386 ES-T- 2059749 US-A- 5093365	17/05/94 09/06/94 16/11/94 03/03/92

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